# Synthesis of Mixed Bisamides of Cyclic ortho-Dicarboxylic Acids 

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#### Abstract

A procedure was developed for preparation of bisamides of cyclic ortho-dicarboxylic acids by acylation of $m$ - or $p$-phenylenediamine in acetone solution at room temperature simultaneously with two different anhydrides of cyclic or aromatic ortho-dicarboxylic acids; another process consisted in treating an anhydride of aromatic or cyclic dicarboxylic acid with monoamide of cis-4-cyclohexene-1,2-dicarboxylic acid in dimethylformamide at room temperature.


Amidoacids from aromatic and cyclic ortho-dicarboxylic acids serve as excellent monomers for preparation of heat-resistant polyamides [1].

Symmetrical bisamides were prepared before [2] by acylation of aromatic diamines with anhydrides of cyclic ortho-dicarboxylic acids. The ready formation of bisamides shows that acylation of both amino groups in $m$-and $p$-phenylenediamines occurs with similar rates.

We describe here a procedure for preparation of mixed bisamides where the ratio of reactants,
mixing order, and character of solvent play significant role.

A simultaneous mixing of acetone solutions containing equimolar quantities of $m$ - or $p$-phenylenediamine (II, III) anhydride of cis-4-cyclohexene-1,2dicarboxylic acid (I), and also anhydrides of phthalic (IV), trans-4,5-dibromocyclohexane-1,2-dicarboxylic (V), 1,4,5,6,7,7-hexachloro[2.2.1]hept-5-ene-2,3-dicarboxylic (VI) or 1,2,3,4,11,11-hexachlorotricyclo[6.2.1. ${ }^{5,10}$ ]undec-2-ene-7,8-dicarboxylic (VII) acids in acetone medium at room temperature results in formation of mixed bisdiamides X-XVII.

Characteristics and elemental analyses of bisamides X-XVII and bisimides XVII-XXV

| Compd. no. | Yield, \% | $\mathrm{mp},{ }^{\circ} \mathrm{C}$ (from benzene) | $R_{\text {f }}$ | Found, \% |  |  |  | Formula | Calculated, \% |  |  |  | M |  | Acid number |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | C | H | Hlg | N |  | C | H | HIg | N | found | calcd. | found | calcd. |
| X | 97 | 299 | 0.39 | 64.38 | 4.65 | - | 6.60 | $\mathrm{C}_{22} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{6}$ | 64.70 | 4.90 | - | 6.86 | 405.5 | 408.0 | 270.1 | 275.8 |
| XI | 99 | 315 | 0.41 | 46.00 | 4.02 | 27.26 | 4.82 | $\mathrm{C}_{22} \mathrm{H}_{24} \mathrm{Br}_{2} \mathrm{~N}_{2} \mathrm{O}_{6}$ | 46.15 | 4.20 | 27.97 | 4.90 | 566.6 | 572.0 | 196.8 | 197.8 |
| XII | 97 | 321 | 0.46 | 43.58 | 2.48 | 33.20 | 4.00 | $\mathrm{C}_{23} \mathrm{H}_{18} \mathrm{Cl}_{6} \mathrm{~N}_{2} \mathrm{O}_{6}$ | 43.34 | 2.85 | 33.75 | 4.44 | 629.5 | 631.0 | 176.8 | 177.5 |
| XIII | 97 | 348 | 0.50 | 47.21 | 3.42 | 30.20 | 4.00 | $\mathrm{C}_{27} \mathrm{H}_{24} \mathrm{Cl}_{6} \mathrm{~N}_{2} \mathrm{O}_{6}$ | 47.30 | 3.50 | 31.9 | 4.00 | 684.2 | 685.0 | 162.4 | 163.3 |
| XIV | 99 | 281 | 0.41 | 65.00 | 4.79 | - | 6.80 | $\mathrm{C}_{22} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{6}$ | 64.08 | 5.83 | - | 6.80 | 405.0 | 408.0 | 274.6 | 275.8 |
| XV | 98 | 301 | 0.42 | 46.00 | 4.03 | 27.27 | 4.60 | $\mathrm{C}_{22} \mathrm{H}_{18} \mathrm{Br}_{2} \mathrm{~N}_{2} \mathrm{O}_{6}$ | 46.15 | 4.20 | 27.97 | 4.90 | 566.8 | 572.0 | 196.8 | 197.8 |
| XVI | 98 | 296 | 0.43 | 43.61 | 2.66 | 33.18 | 4.10 | $\mathrm{C}_{23} \mathrm{H}_{18} \mathrm{Cl}_{6} \mathrm{~N}_{2} \mathrm{O}_{6}$ | 43.74 | 2.85 | 33.75 | 4.44 | 630.0 | 631.0 | 177.0 | 177.5 |
| XVII | 97 | 334 | 0.51 | 47.21 | 3.48 | 30.88 | 4.00 | $\mathrm{C}_{27} \mathrm{H}_{24} \mathrm{Cl}_{6} \mathrm{~N}_{2} \mathrm{O}_{6}$ | 47.30 | 3.50 | 31.09 | 4.09 | 684.0 | 685.0 | 162.8 | 163.5 |
| XVIII | 94 | 306 | 0.69 | 70.08 | 4.00 | - | 7.21 | $\mathrm{C}_{22} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{4}$ | 70.97 | 4.30 | - | 7.53 | 370.2 | 372.0 |  |  |
| XIX | 96 | 220 | 0.58 | 49.33 | 3.00 | 29.71 | 4.98 | $\mathrm{C}_{22} \mathrm{H}_{16} \mathrm{Br}_{2} \mathrm{~N}_{2} \mathrm{O}_{4}$ | 49.62 | 3.61 | 30.10 | 5.26 | 531.0 | 532.0 |  |  |
| XX | 96 | 230 | 0.70 | 46.01 | 2.10 | 35.61 | 4.44 | $\mathrm{C}_{23} \mathrm{H}_{14} \mathrm{Cl}_{6} \mathrm{~N}_{2} \mathrm{O}_{4}$ | 46.38 | 2.35 | 35.80 | 4.70 | 370.0 | 372.0 |  |  |
| XXI | 95 | 306 | 0.69 | 49.25 | 2.89 | 32.41 | 4.01 | $\mathrm{C}_{27} \mathrm{H}_{20} \mathrm{Cl}_{6} \mathrm{~N}_{2} \mathrm{O}_{4}$ | 49.92 | 3.08 | 32.82 | 4.31 | 647.8 | 649.0 |  |  |
| XXII | 79 | 230 | 0.73 | 70.51 | 3.92 | - | 7.04 | $\mathrm{C}_{22} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{4}$ | 70.97 | 4.30 | - | 7.53 | 370.2 | 372.0 |  |  |
| XXIII | 99 | 194 | 0.67 | 49.07 | 2.71 | 29.22 | 4.88 | $\mathrm{C}_{22} \mathrm{H}_{16} \mathrm{Br}_{2} \mathrm{~N}_{2} \mathrm{O}_{4}$ | 49.62 | 3.01 | 30.10 | 5.26 | 528.4 | 532.0 |  |  |
| XXIV | 96 | 202 | 0.68 | 45.91 | 2.11 | 35.48 | 4.32 | $\mathrm{C}_{23} \mathrm{H}_{14} \mathrm{Cl}_{6} \mathrm{~N}_{2} \mathrm{O}_{4}$ | 46.38 | 2.35 | 35.80 | 4.70 | 594.4 | 595.0 |  |  |
| XxV | 88 | 308 | 0.71 | 49.69 | 3.21 | 32.66 | 4.14 | $\mathrm{C}_{27} \mathrm{H}_{16} \mathrm{Cl}_{6} \mathrm{~N}_{2} \mathrm{O}_{4}$ | 50.23 | 2.48 | 33.02 | 4.34 | 643.6 | 645.0 |  |  |

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Bisamides X-XVII were also obtained by reaction of monoamides VIII or IX with anhydrides IV-VI or VII in DMF at room temperature. Since monoamides VIII, IX are insoluble in acetone, their reaction with anhydrides IV-VI or VII occurs only in DMF that ensures homogeneity of the medium. The developed procedure provides a possibility to obtain any mixed bisamides by acylation of $m$ - or $p$-phenylenediamine varying the order of anhydride addition and the solvent.

The proof of the structure of mixed bisamides X-XVII was performed by transformation thereof into bisimides XVIII-XXV by heating the respective amides in DMF for $5-6 \mathrm{~h}$ at boiling ( $152^{\circ}$ ). The characteristics of bisamides $\mathbf{X}$-XVII and bisimides XVIII-XXV are given in the table.

The structure of the mixed bisamides is also confirmed by IR spectra containing the stretching bands of $\mathrm{C}=\mathrm{C}$ bonds at $1660,1610 \mathrm{~cm}^{-1}$ and of amide group at $3290,3220 \mathrm{~cm}^{-1}$. The stretching vibrations of the carboxy group appear as a strong shoulder at $1720 \mathrm{~cm}^{-1}$. The band at $2600 \mathrm{~cm}^{-1}$ indicates that the carboxy group is linked to the adjacent amide group by an intramolecular hydrogen bond. The presence of carboxy group in compounds X-XVII was also proved by measuring the acid numbers.

## EXPERIMENTAL

IR spectra of bisamides X-XVII were recorded on spectrophotometer UR-20 from mulls in mineral oil [3]. The acid numbers were determined by potentiometric titration with the use of LPM-60M device by procedure [4]. The purity of compounds was checked by TLC on silica gel of L5/40 $\mu$ grade, eluent benzene-dichloroethane-acetic acid, 4:1.5:1 (by volume), visualizing of spots under UV irradiation [5]. Molecular weight was determined by cryoscopy in camphor [6].

Anhydride of cis-4-cyclohexene-1,2-dicarboxylic acid was obtained as in [7], anhydride of trans-4,5-di-bromocyclohexane-1,2-dicarboxylic along procedure [8], anhydride of 1,4,5,6,7,7-hexachloro[2.2.1]hept-5-ene-2,3-dicarboxylic acid was prepared by method [9], anhydride of $1,2,3,4,11,11$-hexachlorotricyclo[6.2.1.0 ${ }^{5,10}$ ]undec-2-ene-7,8-dicarboxylic acid by method [10].

Synthesis of cis-4-cyclohexene-1,2-dicarboxylic acid monoamides VIII, IX. To a solution of 0.1 mol of anhydride $\mathbf{I}$ in 20 ml of acetone was added at vigorous stirring a solution of 0.1 mol of meta-(II) or para-(IV) phenylenediamine in 100 ml of acetone.

The separated crystals were filtered off, washed with acetone, and dried.
cis-4-Cyclohexene-1,2-dicarboxylic acid $N$-(maminophenyl)amide (VIII). Yield $99 \%$. mp $189^{\circ} \mathrm{C}$ (from benzene). $R_{\mathrm{f}} 0.30$. Found, \%: C 64.05; H 5.91 ; N 10.21. $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3}$. Calculated, \%: C 64.62; H 6.15; N 10.77.
cis-4-Cyclohexene-1,2-dicarboxylic acid $N$-( $p$ aminophenyl)amide (IX). Yield $98 \%$. mp $202^{\circ} \mathrm{C}$ (from benzene). $R_{\mathrm{f}} 0.31$. Found, $\%: \mathrm{C} 63.91$; H 5.88 ; N 10.29. $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3}$. Calculated, \%: C 64.62; H 6.15; N 10.77.

Synthesis of mixed bisamides X-XVII. (a) To a solution of 0.1 mol of anhydride (I) and 0.1 mol of anhydride IV-VI or VII in 300 ml of acetone at room temperature while vigorous stirring was added solution of 0.1 mol of meta-(II) or para-(IV) phenylenediamine in 50 ml of acetone. The mixture was left standing for 12 h till crystals precipitated. The precipitate was filtered off, washed with acetone, dried, and analyzed. The characteristics of bisamides $\mathbf{X}$ XVII are given in table.
(b) To a vigorously stirred solution of 0.1 mol of monoamide VIII or IX in 300 ml DMF was gradually added at room temperature 0.1 mol of anhydride IV-VII dissolved in 50 ml of DMF. The mixture was left standing for 12 h . The precipitated crystals were filtered off, dried, and recrystallized from benzene. Yield of bisamides X-XVII 96-99\%. Characteristics of compounds X-XVII prepared along procedures (a) and (b) are identical.

Synthesis of mixed bisimides XVIII-XXV. A mixture of 0.1 mol of mixed bisamide X-XVII and 200 ml of DMF was heated to $152^{\circ} \mathrm{C}$ for $5-6 \mathrm{~h}$. The cooled mixture was poured into ice water at stirring. The separated crystals were filtered off, washed with distilled water, and recrystallized from methanol. Yield of bisimides XVIII-XXV 79-96\%, their characteristics are given in table.

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